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A.P. Chemistry Practice

Test: Ch. 11, Solutions

Name _____ MULTIPLE

CHOICE. Choose the

one alternative that

best completes the

statement or answers

the question. 1)

Formation of solutions

where the process is

endothermic can be

spontaneous provided

that _____. A)the

solvent is a gas and

the solute is a solid

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MULTIPLE ...

Chapter 11 AP

Chemistry. gases.

liquids. solids.

converting a gas into a

liquid or solid.... -highly

compressible...

-assumes shape of

volume of container ...

-m.... -particles are not

spread out ... -almost

incompressible ...

-assume....

-incompressible and

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have definite shape
and volume ...
-molecules....

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(Chapter 11) Multiple
Choice (50%) Please
use the following
choices to answer
questions 1-7. A)
London dispersion
forces B) Ion-ion
attractions C) Dipole-

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dipole attractions D)
Dipole-ion attractions
E) Hydrogen bonding
H) Dipole-induced
dipole attractions I) Ion-
induced dipole
attractions ...

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AP Chemistry Chapter
11. Intermolecular
Forces, Liquids, and
Solids - 10 - Vapor
Pressure and Boiling
Point • Liquids boil
when the external

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pressure at the liquid surface equals the vapor pressure. • The normal boiling point is the boiling point at 760 mm Hg (1 atm).

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of Solutions . 11.1

Solution Composition .

A. Molarity 1. liters of
solution moles solute

Molarity(M) = B. Mass
Percent 1. $\times 100 =$

mass of. solution mass
of solute Mass percent.

C. Mole Fraction . 1. D.

Molality 1. ki ram of
solvent moles of solute

Molality log = E.

Normality 1. liter of
solution equivalents

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Properties of Solutions

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Key. Key Solutions

Concentration. Chapter

11 Practice Key . Ch

12: Ch 12 Practice

Answers . Ch 13:

Equilibrium Handout

Key . Equilibrium

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19: Ch 14 Ch19

Practice Keys .

Baker, Mrs.

(Science) / AP

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11%-15% of exam

score. Unit 9: Practice Test

Applications of

Thermodynamics You'll

be introduced to the

concept of

"thermodynamic

favorability" for

reactions, meaning

how likely they are to

occur given energy

changes and

environmental factors.

Topics may include: ...

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transformations; and
energy.

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Solutions. Formula ...

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due date Points 38;
Questions 38; Time
Limit None Allowed
Attempts 2 Take the
Quiz.

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At a given temperature
the vapor pressure of
pure liquid diethyl
ether and isopropyl
alcohol are 730 torr

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Chapter 11 and 310 torr,

respectively. A solution prepared by mixing isopropyl alcohol and diethyl ether obeys Raoult's law.

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difficulty as review
them wherever you are
now. Ap Chemistry

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Test AP Chemistry:

Intermolecular Forces,
Liquids, and Solids

Lecture Outline 11.1 A
Molecular Comparison
of Liquids and Solids

Physical properties of
liquids and solids are
due to intermolecular
forces.

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