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coefficients.) (a) The
number of atoms in
state 1 can increase
due to spontaneous
decay from state 2,
and increase due to
stimulated emission
from state 2, and
decrease due to
stimulated absorption
up to state 2. The rate
of each of these
processes is equal to
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A lithium nucleus has
four independent spin
orientations,
conventionally labeled
by the quantum
number $m = \pm 3/2, \pm 1/2,$
 $1/2, 3/2$. In a magnetic
field B , the energies of

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these four states are $E = \pm \mu B$, where the constant μ , is $1.03 \times 10^{-7} \text{ eV} / \text{T}$. In the Purcell-Pound experiment described in Section 3.3, the maximum

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book since material from chapters 7 and * will be covered later in Phys 328. We will cover tentatively chapters 1-6 at an average pace of roughly 10 textbook pages per class, and if time will allow I will try to cover as well material from chapter 7 or even chapter 8.

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